

# Sulfur Content and Isotopic Composition of Lichen Species Bullion Mine; Basin, MT

Jade A. Marks<sup>1</sup>, Lisa M. Pratt<sup>2</sup>, Seth A. Young<sup>2</sup>

<sup>1</sup> BSES Program, Indiana University

<sup>2</sup> Department of Geological Sciences, Indiana University

# Background

- Lichens are useful indicators of environmental impact.

## • Variables Of Interest

- Stable isotopic composition of carbon and sulfur
- Species richness and distribution
- Heavy metal concentrations



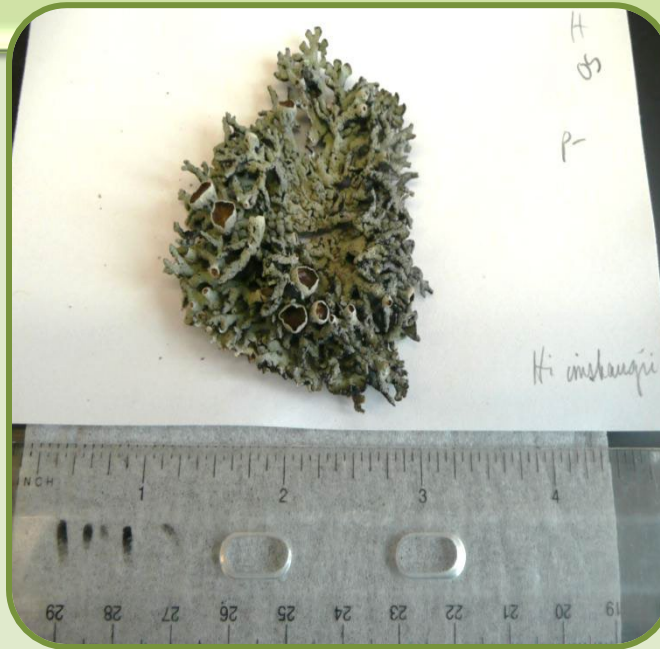
Lichen specimens divided by species during identification procedure at Montana State University, July 2011.

- Established methods for assessing the sources and effects of air pollution.



# Abstract

Photograph of lichen species *Hypogymnia imshaugii*. Scale in inches.



Detailed photograph of lichen species *Pelliqua caninia*.

- Sulfur concentration and isotopic signature in lichen species
- Abandoned precious metal mine in central Montana
- Specimens analyzed for heavy metal concentrations:
  - Lead (Pb)
  - Chromium (Cr)
  - Copper (Cu)
  - Cadmium (Cd)
  - Zinc (Zn)
- Evaluate the potential of lichens to assess human disturbance in remote mountainous areas.

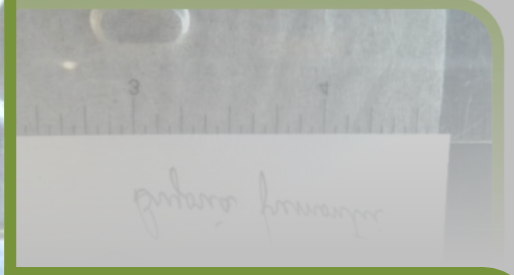
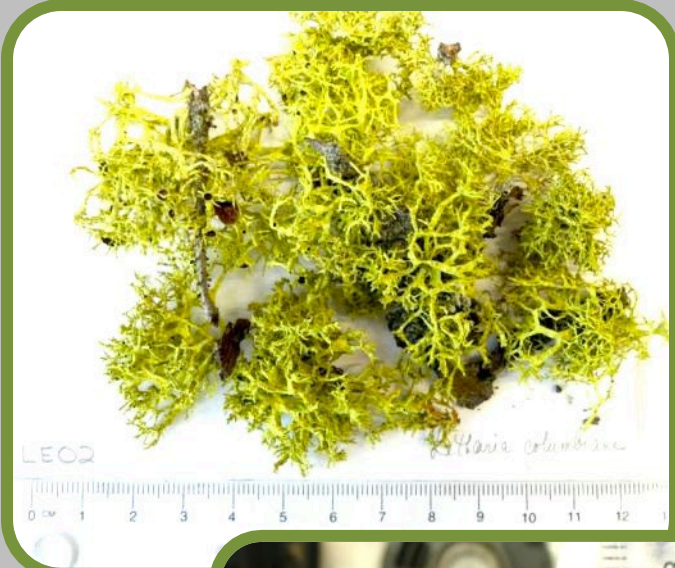
# Research Questions

- Can lichens indicate impacts of remote mining operations?
- Are heavy metals present in detectable quantities?
- How can we best isolate sulfur from lichens?
- Do concentrations of sulfur and metals vary spatially or among species/genera?
- Does distance from mine correlate with concentrations?





# What is a lichen?





A lichen is . . .

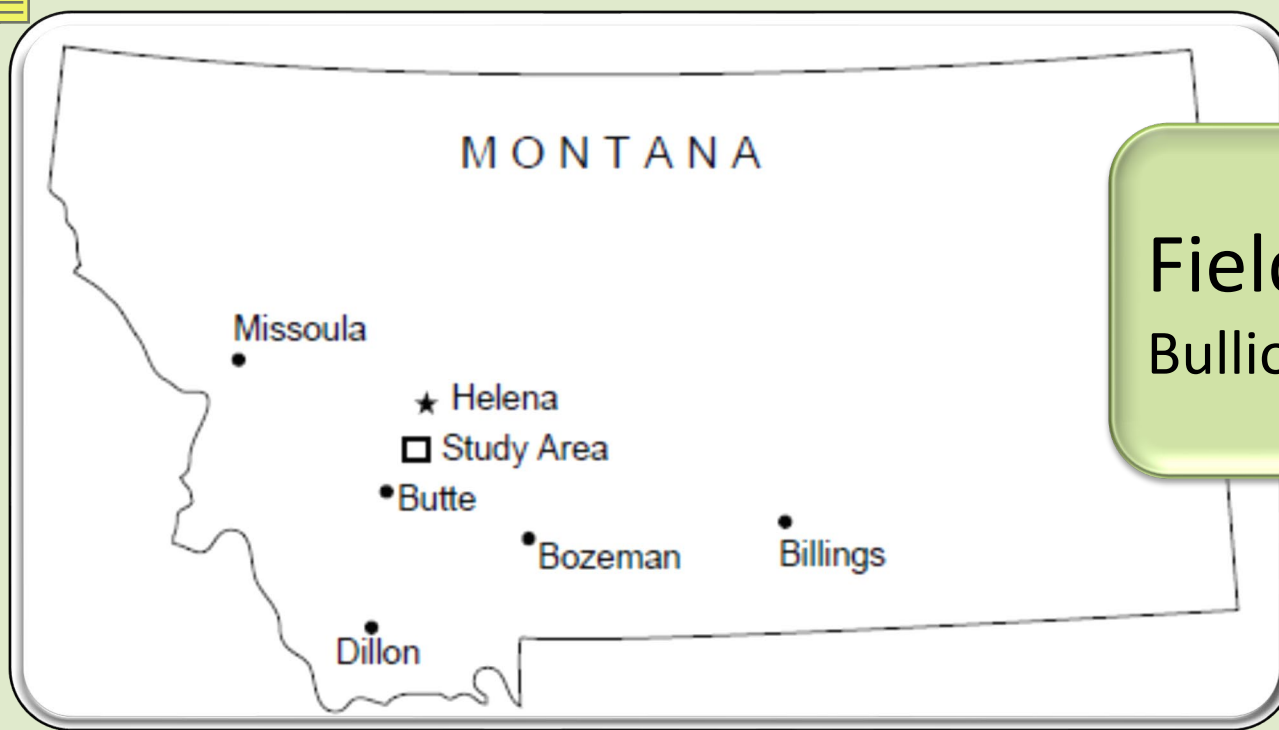
Not a plant

Categorized as a *cryptogam*

Symbiotic partnership between a photobiont and a fungus

An evolutionary success: over 14,000 species

Diverse in size, growth form, and color



**Field Site:**  
Bullion Mine; Basin, MT

- Basin Mining District; Jefferson County, Montana.
- Mined for gold and silver between 1897 and 1955.
- Superfund National Priorities List site in 1999.
- Ore body contains metallic sulfides:
  - Pyrite ( $\text{FeS}_2$ )
  - Galena ( $\text{PbS}$ )
  - Sphalerite ( $(\text{Zn,Fe})\text{S}$ )
  - Tetrahedrite  $(\text{Cu,Fe})_{12}\text{Sb}_4\text{S}_{13}$
  - Arsenopyrite ( $\text{FeAsS}$ )
- Initial cleanup completed in 2002; still actively monitored by the EPA.



# Field Methods

- Twenty-six samples consisting of six lichen species.
- Eleven locations at distances of 40-250 m. from the center of Bullion
- Sampling conducted in a roughly radial pattern, as allowed by disturbance and abundance.
- Samples stored in paper envelopes in a cool, dry, dark place.
- Water samples of acid mine drainage, preserved with cadmium chloride, collected with basic field chemistry

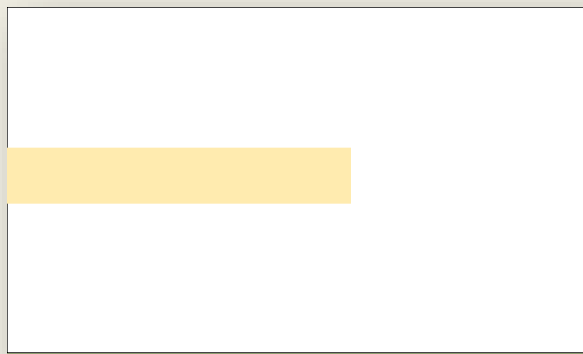


Bullion Mine, Jefferson County. Basin, Montana

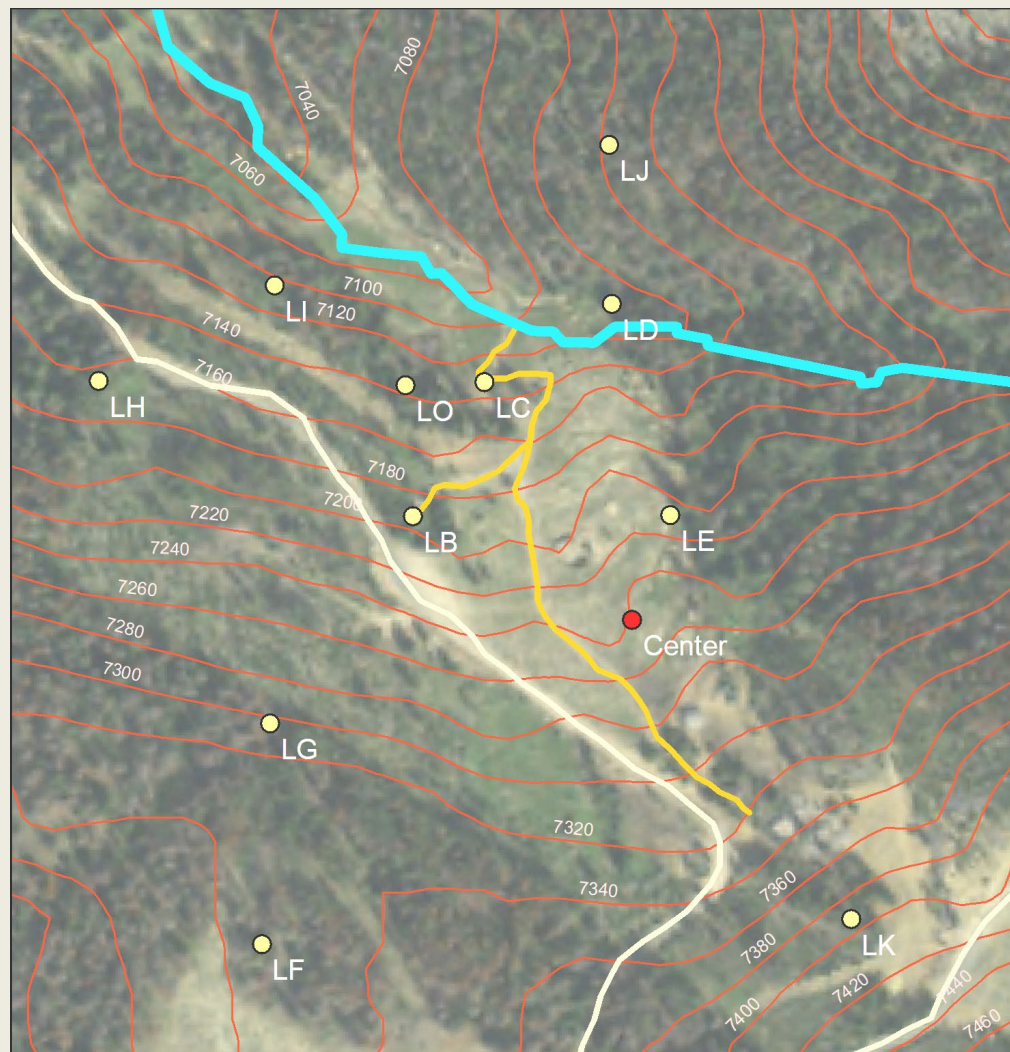


# Lichen Sample Sites

Bullion Mine; Basin, MT  
July 2011



Sample Site	Dist. from Cntr (m)
LB	92.05
LC	111.25
LD	125.27
LE	42.37
LF	193.12
LG	146.91
LH	241.40
LI	193.12
LJ	193.12
LK	146.91
LO	137.90



# Lab Methods

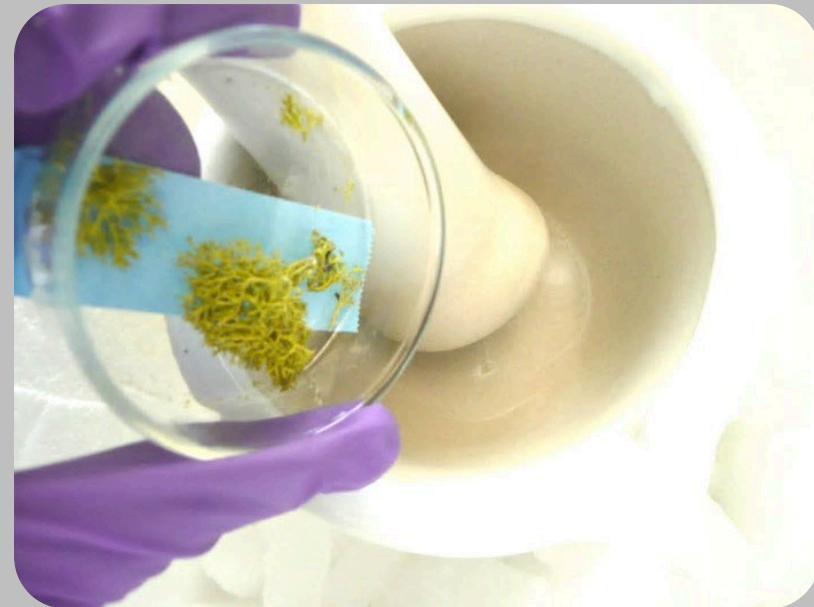
2. Unpulverized specimens analyzed for preliminary C/S ratios



Eltra CS-2000 Carbon-Sulfur Determinator

3. Lichens pulverized by freezing with liquid nitrogen and grinding into a fine powder.

1. Lichen identification
  - chemical tests
  - investigate spore structure under a dissecting microscope
  - examine structure of fruiting bodies.



Whole lichen prepared for pulverization



# Lab Methods: continued

Begin with chemically-untreated lichen powder

**1**

Aliquots (15 mg) of lichen powder taken randomly, whenever necessary

•  
Continuous flow-isotope ratio mass spectrometry (CF-lichen)

**2**

30 (1 g) aliquots of lichen powder

•  
BaSO<sub>4</sub> precipitated by Parr Bomb™ oxidation

•  
Continuous flow-isotope ratio mass spectrometry (CF-BaSO<sub>4</sub>)

**3**

4 (4 g) aliquots of lichen powder

•  
BaSO<sub>4</sub> precipitated by Parr Bomb™ oxidation

•  
Mixed BaSO<sub>4</sub>

•  
SO<sub>2</sub> prepared from 5 aliquots of BaSO<sub>4</sub>, by SO<sub>2</sub> extraction line techniques

•  
Dual-inlet isotope ratio mass spectrometry (DI-IRMS)

**4**

3 (1g) aliquots of lichen powder

•  
Rinse sample with 400 ml Milli-Q water to separate H<sub>2</sub>O-soluble SO<sub>4</sub>

•  
Isolate elemental S fraction (soxhlet extraction, 24 hrs with CH<sub>2</sub>Cl)

•  
Concentrate extracted CH<sub>2</sub>Cl, Parr Bomb™ for volatile organo-sulfur

•  
Acid volatile S extraction (2 hrs in 6 N HCl with N<sub>2</sub> headspace)

•  
Isolate chrome reducible sulfur (0.2 M CrCl<sub>2</sub>, 2 hrs in 12 N HCl with N<sub>2</sub> headspace)

•  
Parr Bomb™ residue for acid resistant organic sulfur compounds

•  
All fractions precipitated as BaSO<sub>4</sub>, and analyzed by continuous flow-isotope mass spectrometry

Four methods for extracting S from the lichen thallus. Methods 1-3 are described in Yun (2004). Method 4 is modified from Lefticariu (2006). Methods 1, 2, and 4 are used in this study. Figure adapted from Yun 2004.

## Lab Methods: continued

- ❑ Ground specimens were freeze dried before extraction.

- Water-soluble sulfur
- Elemental sulfur
- Acid-volatile sulfur
- Acid-soluble sulfate
- Chrome-reducible sulfides
- Solvent-soluble organosulfur

- ❑ Three samples selected for sequential extraction.



Freeze-drying station



Soxhlet extraction setup for elemental S



Extraction line for acid volatile S



## Lab Methods: continued



Parr® Oxygen Bomb

- ❑ Sulfur collected as silver sulfide or barium sulfate

- ❑ Twenty-three samples oxidatively combusted to liberate sulfur within the thallus.

- ❑ Analyzed for stable-isotopic composition using a Finnegan MAT 252 isotope ratio mass spectrometer



AAlylist 800 AAS

- ❑ Eight samples analyzed for heavy metals by atomic absorption spectroscopy

# Results

**Table 1**

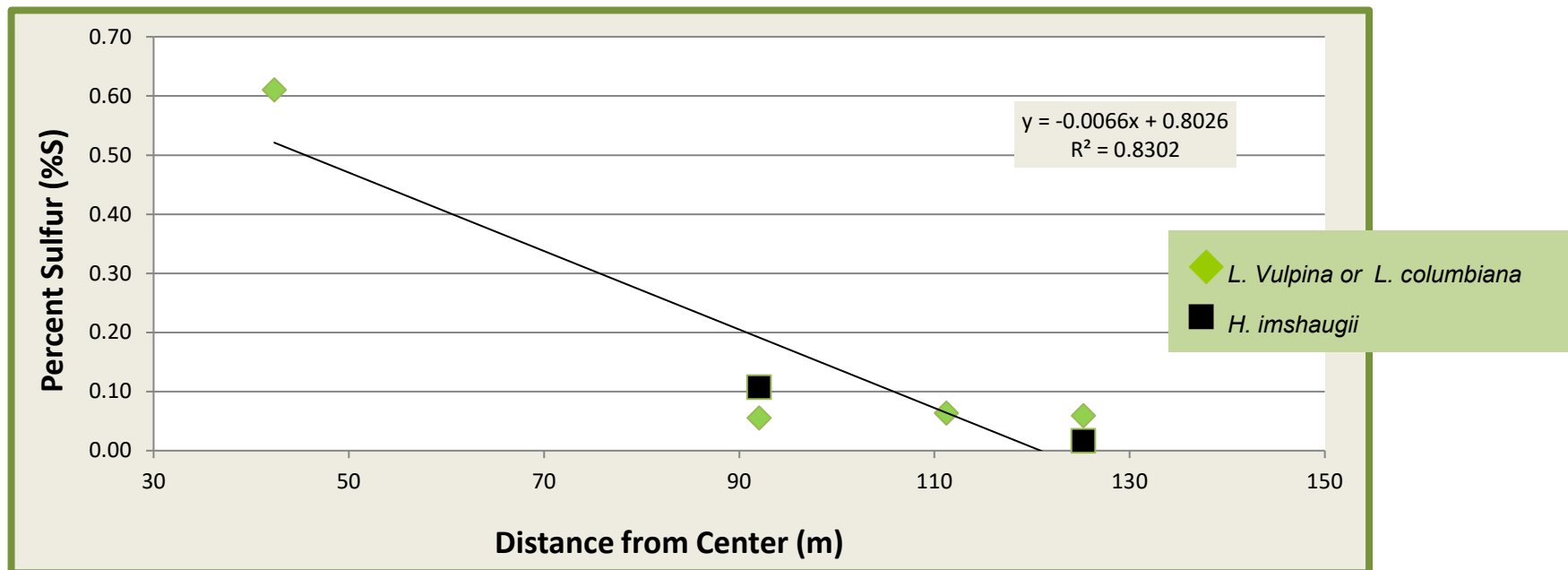
Percent Sulfur by C/S Determinator

Lichen	%S
<i>Hypogymnia</i> sp.	0.130
<i>Letharia</i> sp.	0.204
<i>Letharia</i> sp.	0.251
<i>Letharia</i> sp.	0.215
<i>Letharia</i> sp.	0.158
<i>Letharia columbiana</i>	0.138

**Table 2**

Percent Sulfur by Gravimetric Methods

Lichen	Dist. (m)	%S
<i>Hypogymnia imshaugii</i>	92.05	0.107
<i>Hypogymnia imshaugii</i>	125.27	0.017
<i>Letharia vulpina</i>	111.25	0.055
<i>Letharia vulpina</i>	125.27	0.063
<i>Letharia vulpina</i>	42.37	0.610
<i>Letharia columbiana</i>	92.05	0.059





# Results: continued

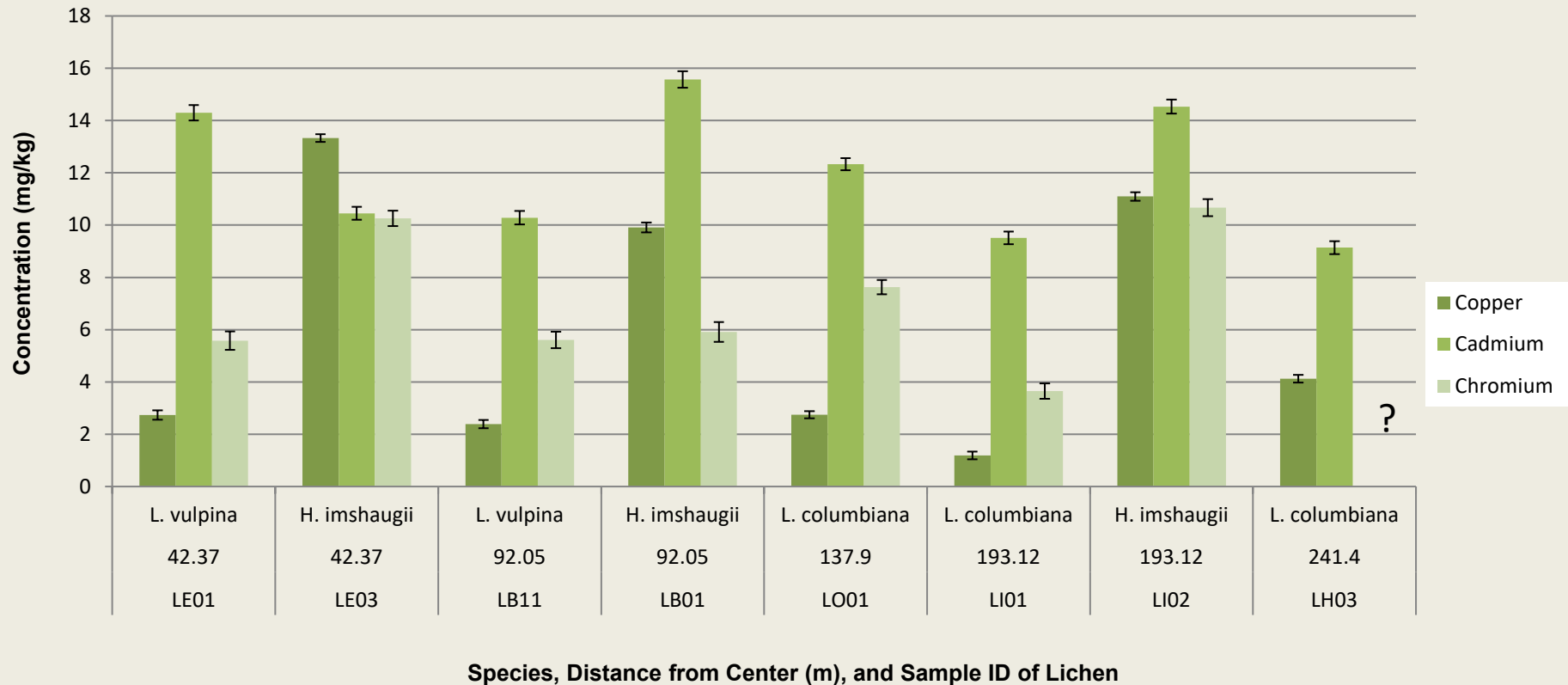


Figure 1: Concentration of copper, cadmium, and chromium (mg/kg or ppm) in eight lichen specimens from five different sample locations.

- No apparent trend between Cu, Cd, or Cr concentrations and radial distance from field site

# Results: continued

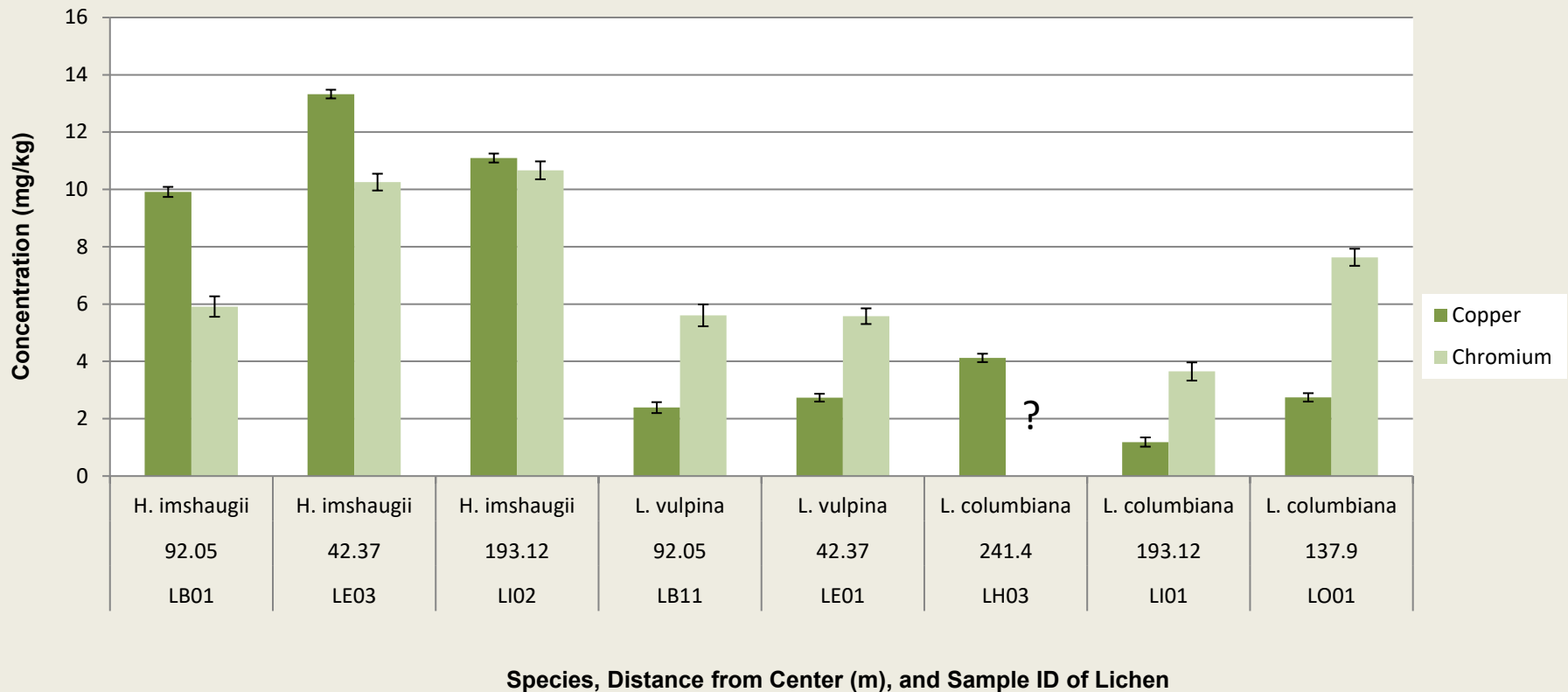


Figure 2: Concentration of copper, and chromium (mg/kg or ppm) in eight lichen specimens from five different sample locations.

- Cu and Cr concentrations appear to vary between lichen species
- *H. imshaugii* exhibits higher concentrations of Cu than both species of *Letharia*, regardless of distance from the center of the mine.
- Inversion of Cr and Cu as the dominant metal between genera

# Results: continued

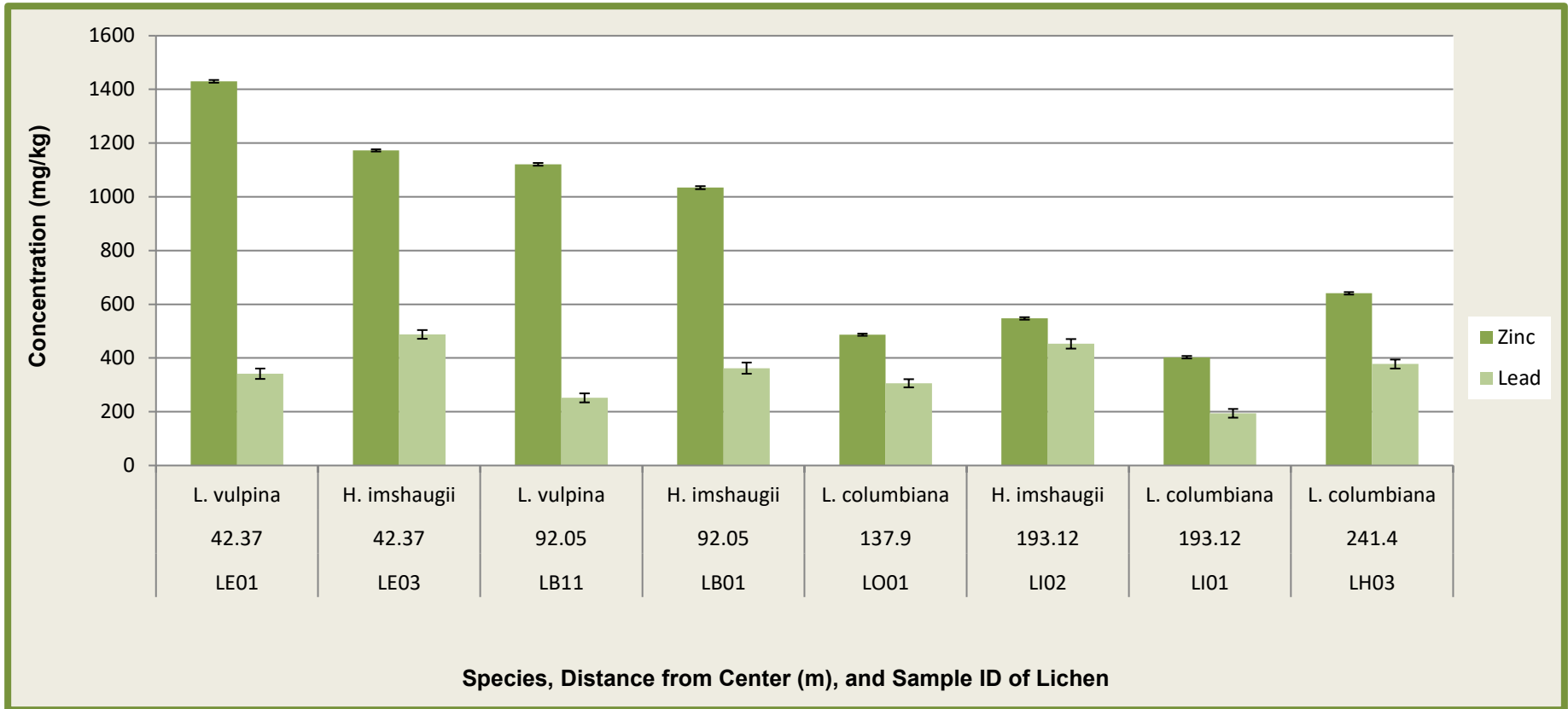


Figure 3: Concentration of zinc and lead (mg/kg or ppm) in eight lichen specimens from five different sample locations.

- Trend of decreasing Zn concentration with increasing radial distance from the center of the mine



# Conclusions



- ❑ Concentrations of sulfur and heavy metals vary
  - Spatially across an area of disturbance
  - Between species occurring in the same location.
  
- ❑ It will be possible to use lichens to assess environmental impact of mining activities in remote mountainous areas, despite the small scale of operation.
  
- ❑ Preliminary data does not confirm the *extent* to which the lichens growing near the Bullion Mine were affected by anthropogenic disturbance.
  
- ❑ Further research:
  - Examine a control specimen
  - Continuing stable-isotope research
  - Conduct a geospatial analysis

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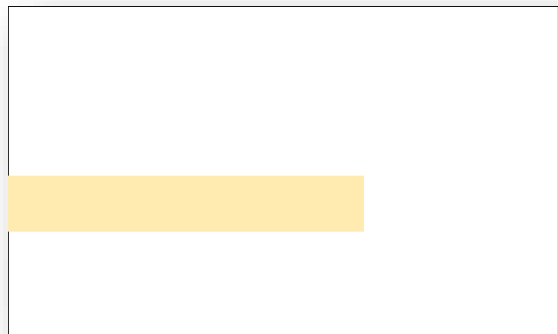
Questions?



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